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# Nitrous oxide and methane transport through rice plants

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Abstract The separate closed chamber technique was used to study the potential of rice plants for transporting N<sub>2</sub>O and CH<sub>4</sub> produced in soil to the atmosphere. The results indicate that N<sub>2</sub>O produced in soil can be conducted to the atmosphere via rice plants similarly to CH<sub>4</sub> transport. More than 80% of both N<sub>2</sub>O and CH<sub>4</sub> was emitted through rice plants. The rest was emitted through the soil/ water/atmosphere interface by ebullition and diffusion. Nitrate addition increased the total N<sub>2</sub>O emission rate substantially but decreased the total CH<sub>4</sub> emission. Nitrate addition did not change the CH<sub>4</sub> emission ratio through rice plants, but lowered the percentage of N<sub>2</sub>O emission through rice plants. The results suggest that rice plants serve not only as a conduit for most of the CH<sub>4</sub> leaving the soil but also for the N<sub>2</sub>O produced in the soil.

Key words Methane  $\cdot$  Nitrous oxide  $\cdot$  Greenhouse gases  $\cdot$  Flooded rice soil

### Introduction

Growing awareness of global warming and ozone destruction has led to an increased emphasis on the study of greenhouse gases such as  $CO_2$ ,  $CH_4$ ,  $N_2O$  and CFCs. Of these,  $CH_4$  and  $N_2O$  are two important trace gases with radiative activities and are next only to  $CO_2$  in their contribution to global warming.

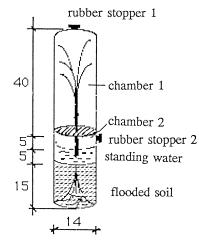
Three processes including ebullition, molecular diffusion and lacunae transport by plants through aerenchyma have been recognized as regulating the flux of  $CH_4$  from paddy fields into the atmosphere (Wang et al. 1994). Of the three pathways, aerenchyma transport is considered the

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Z.P. Wang (▷ ) Wetland Biogeochemistry Institute, Louisiana State University, Baton Rouge, LA 70803-7511, USA most important when rice plants are involved in the ecosystem. Rice plants develop an intercellular gas space system – aerenchyma, which provides the roots with  $O_2$ . This gas space system also enables the transport of other gases, including CH<sub>4</sub> and CO<sub>2</sub> from the soil/sediment to the atmosphere. The transport of CH<sub>4</sub> via rice plants has been well documented by many authors (Cicerone and Shetter 1983; Seiler et al. 1984; Inubushi et al. 1990; Bouwman 1990). Most of the knowledge on the issue of gas transport by plants has obtained with CH<sub>4</sub>. Few studies of N<sub>2</sub>O emission through aquatic plants have been conducted. Mosier et al. (1990) and Bhadrachalam et al. (1992) found the fluxes of N gases from the soil were greater when rice plants were enclosed in the system. This suggests that N<sub>2</sub>O produced in the flooded soil system may also be transported through rice plants. To test this hypothesis, we applied the separate closed chamber method to quantify the amounts of N<sub>2</sub>O, in comparison with CH<sub>4</sub>, emitted from plants and soil separately.

## Materials and methods

The experiment was conducted in Shenyang Experimental Station of Ecology, Chinese Academy of Sciences (41°32'N, 122°23'E). The soil used was a meadow brown paddy soil (silt loam) with an organic matter content 16.17 g kg  $^{-1},$  total N 0.76 g kg  $^{-1}$  and pH 6.4, respectively. Three rice plants including whole roots at the booting stage were dug out from the field with soil and transplanted into growth chambers maintained with 5 cm standing water. The separate closed chamber used in this study is shown in Fig. 1. It consists of two gas collection chambers for gas emission from shoot and soil separately. The PVC soil container was 14 cm in diameter and 25 cm in height. It was filled with soil and standing water up to 20 cm and the remaining 5 cm was left for collection of gases emitted from soil. The gas possibly emitted from the 10-cm base part of sheaths was included in the emission from soil/water in this experiment. The top cover chamber was made of transparent Plexiglas and was 40 cm in height. The two chambers were separated by two pieces of half-circle PVC plates which had a center hole (about 0.5 cm in diameter) to allow the penetration of rice plants. The rice plant was carefully sealed with silicon sealant to prevent gas leaks between the two chambers. Rubber stoppers were installed at the top of the upper chamber and at the side of the lower chamber to allow gas sampling. Methane measurement was



**Fig. 1** Schematic diagram of the closed chamber system method (units are centimeters)

taken 1 week after re-transplanting to allow the rice plants to adopt to the new growth environment. For each measurement, the top chamber was placed of the top of the container and sealed with silicone sealant for 24 h. Gas samples in the headspace of both chambers were taken every 4 h by syringe and transferred into a 15-ml Vacutainer immediately.

The effect of nitrate N application on  $N_2O$  emission followed the same sampling procedure. One hundred milligrams  $KNO_3$ -N solution was added to the water phase after  $CH_4$  measurement. Two days later, the chambers were sealed again and the gas collection procedure was repeated.

Both CH<sub>4</sub> and N<sub>2</sub>O concentrations in the Vacutainers were analyzed by gas chromatography. For CH<sub>4</sub> analysis, an SP-2305E GC (Gas chromatography) equipped with FID (Flame ionization detector) was used; detector, oven and injector temperatures were 170, 140 and

Table 1 Percentage of  $CH_4$  and  $N_2O$  emission through rice plants and the soil/water interface

Treatment	$CH_4$		N <sub>2</sub> O	
	Plant	Soil/water	Plant	Soil/water
Control	83.8	16.3	85.7	14.3
N-amended	(17.0) 83.0	(4.0) 17.0	(13.2) 74.6	(2.8) 24.5
	(10.2)	(4.4)	(9.0)	(4.7)

Data in parentheses are standard deviations (n=3)

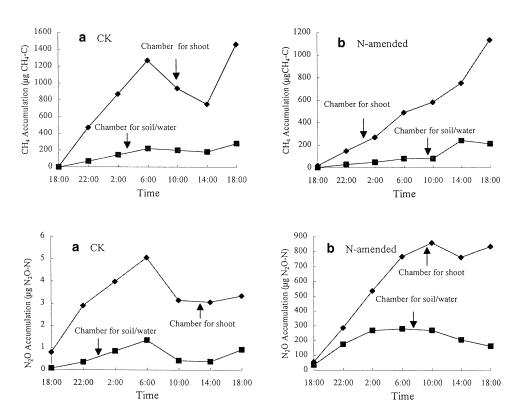
100 °C, respectively. A Shimadzu GC-14A equipped with ECD (Electron capture detector) was used for N<sub>2</sub>O analysis; detector, oven and injector temperatures were 300, 60 and 100 °C, respectively.

#### **Results and discussion**

As shown in Table 1 and Fig. 2, our results for  $CH_4$  agreed well with published data (Schütz et al. 1989; Holzapfel-Pschorn et al. 1985; Sass et al. 1990). Most of the  $CH_4$  produced in the soil was emitted through the rice plant, representing 84% of the total flux. The remaining 16% was emitted through the soil/water interface, although this percentage might be even smaller because the gas possibly emitted from the 10-cm base part of the sheaths was included in the emission from soil/water in this experiment. The ratio was not affected by the decrease in total emission caused by nitrate amendment.

**Fig. 2a, b** Emission of  $CH_4$  through rice plants and the soil/ water interface in a diurnal measurement (illustrated by the amount of  $CH_4$  cumulated in the chambers as a function of time). **a** Control, **b** N-amended

**Fig. 3a, b** Emission of  $N_2O$  through rice plants and the soil/ water interface in a diurnal measurement (illustrated by the amount of  $CH_4$  cumulated in the chambers as a function of time). **a** Control, **b** N-amended



Most of the N<sub>2</sub>O produced in the soil was also emitted through the rice plant. A similar emission pattern to that for CH<sub>4</sub> is shown in Fig. 3. The decrease in cumulated N<sub>2</sub>O might suggest re-solubilization of this water-soluble gas. Emissions from the soil/water/atmosphere interface only contribute a minor part of the total flux of both gases. Long-term flooded paddy field usually shows low denitrification activities due to the limitation of available nitrate. The processes of denitrification and N<sub>2</sub>O emission were greatly stimulated by nitrate addition in our study. Nitrate-amended treatment increased the N<sub>2</sub>O emission substantially 2 days after KNO<sub>3</sub> addition. However, nitrate-N decreased the percentage of N<sub>2</sub>O emission through rice plants from 86% in the control to 75%.

Gas transport is a physical process. Therefore, N<sub>2</sub>O produced in the soil could possibly escape in the same pathway as CH<sub>4</sub>. However, N<sub>2</sub>O is a more water soluble gas with a solubility in water about 20 times that of CH<sub>4</sub> (0.675 and 0.033 g  $l^{-1}$  H<sub>2</sub>O at 20 °C, respectively). This water may act as a carrier of N<sub>2</sub>O, and emission through diffusion might be enhanced for N<sub>2</sub>O. In addition, in combination with the physiological movement of water, such as transpiration, N<sub>2</sub>O may be emitted to the atmosphere through the rice plants. The opportunities for emission of N<sub>2</sub>O through ebullition should be smaller than for CH<sub>4</sub> due to the high solubility of this gas. Further study needs to be conducted to confirm this observation.

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